

EFFECT OF Y_2O_3 AND ZrO_2 ON THE MICROSTRUCTURE AND MECHANICAL PROPERTIES OF NANO-ODS 21Cr-9Mn-6Ni STEELS

VPLIV Y_2O_3 IN ZrO_2 NA MIKROSTRUKTURO IN MEHANSKE LASTNOSTI NANO-ODS 21Cr-9Mn-6Ni JEKEL

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In this article, nano-oxide dispersion-strengthened (ODS) 21Cr-9Mn-6Ni steels were prepared from elemental powders through vacuum hot pressing. Three ODS steels (21Cr-9Mn-6Ni-0.4Ti-0.3 Y_2O_3 , 21Cr-9Mn-6Ni-0.4Ti-0.3 ZrO_2 and 21Cr-9Mn-6Ni-0.4Ti-0.15 Y_2O_3 -0.15 ZrO_2) were prepared to study the effect of Y_2O_3 and ZrO_2 additions on the microstructure and mechanical properties. The nano-oxide precipitates within the austenite matrix were identified using a transmission-electron microscope equipped with energy-dispersive X-ray spectrometer (TEM-EDX). An addition of Y_2O_3 with ZrO_2 leads to the formation of a bimodal distribution of grains and fine oxide particles of Y-Zr-O and Y-Ti-O in the microstructure of an ODS steel. The results of the tensile test at different temperatures reveal that the strength and elongation of 21Cr-9Mn-6Ni steels are improved with the addition of Y_2O_3 and ZrO_2 due to a stronger interface bonding between large numbers of oxide particles and the austenite matrix.

Keywords: ODS steel, nano-oxide precipitates, tensile strength, elongation, fracture surface

V članku avtorji opisujejo izdelavo z nano-oksidi delci disperzijsko utrjenega (ODS) jekla 21Cr-9Mn-6Ni. ODS jeklo je bilo izdelano z vročim stiskanjem iz elementarnih prahov. Tri ODS jekla (21Cr-9Mn-6Ni-0,4Ti-0,3 Y_2O_3 , 21Cr-9Mn-6Ni-0,4Ti-0,3 ZrO_2 in 21Cr-9Mn-6Ni-0,4Ti-0,15 Y_2O_3 -0,15 ZrO_2) so izdelali z namenom, da bi raziskali vpliv dodatkov Y_2O_3 in ZrO_2 na mikrostrukturo in mehanske lastnosti. Nano-oksidi delci v avstenitu so okarakterizirali s presevnim elektronskim mikroskopom, opremljenim z rentgenskim energijskim disperzijskim spektrometrom (TEM-EDX). Ugotovili so, da dodatek Y_2O_3 s ZrO_2 vodi do tvorbe bimodalne porazdelitve kristalnih zrn in drobnih oksidnih delcev Y-Zr-O in Y-Ti-O v mikrostrukturi ODS jekla. Natezni preizkusi pri različnih temperaturah so pokazali, da sta se trdnost in raztezek 21Cr-9Mn-6Ni jekla izboljšala z dodatkom Y_2O_3 in ZrO_2 zaradi močnejše vezi med velikim številom oksidnih delcev in avstenitno matrico.

Keywords: ODS jeklo, nano-oksidi delci, natezna trdnost, raztezek, prelomna površina

1 INTRODUCTION

21Cr-9Mn-6Ni (Nitronic-40) stainless steel is a high-alloyed, high-manganese (Mn) nitrogen (N) strengthened austenitic stainless steel. It has excellent corrosion and oxidation resistance because of a higher amount of chromium (20–30 %) and an increased amount of manganese. Therefore, this steel can be used at elevated temperatures due to its better creep resistance and high-temperature corrosion resistance.¹

The structural components used in the nuclear industry are subjected to a very high temperature and pressure. Stainless steel of different types, including austenitic steel (AS), martensitic-ferritic steel (M-F/S) and super alloys (nickel based) have good high-temperature properties but poor swelling characteristics.² Oxide dispersion strengthened (ODS) steels are the most promising structural materials for nuclear reactors due to many reasons.³ An ODS steel is manufactured using the

dispersion-strengthening mechanism, wherein nano-oxide precipitates are uniformly dispersed in the microstructure.⁴

The superior characteristics of ODS steels depend on the preparation, processing methods⁵ and composition design.⁶ Y_2O_3 nanoparticles with a cubic structure were used as the oxide particles in the development of ODS steels in recent years.⁷ The titanium element was used, along with Y_2O_3 , to optimize the nano-particles in the microstructure of ODS steels.⁸ Researchers also focused on modifying the microstructure of ODS steels by adding zirconium.⁹

In this paper, the effect of Y_2O_3 and ZrO_2 on the development of the ODS 21Cr-9Mn-6Ni steel was investigated. Also, the microstructure and tensile properties (yield strength, ultimate tensile strength and ductility) of three different ODS steels were studied and compared.

2 EXPERIMENTAL PART

Elemental powders of Fe, Cr, Mn, Ni, Si, Mn₃N₂, Ti and C with a purity of above 99.7 % and a size of 45 μm, delivered from Innomet Powders Pvt. Ltd, India, were used as the starting materials. Different ODS materials such as Y₂O₃ and ZrO₂ powders with a purity of above 99.9 % and a size of 500 nm, from Alfa Aesar, UK, were added to the elemental powders to obtain different ODS steels. The powder blends of three different ODS steel compositions and their designations are listed in **Table 1**.

Table 1: Alloy compositions and designations

S. No.	Composition	Alloy designation
1	Fe-21Cr-9Mn-6Ni-Si-0.15N-0.08C-0.4Ti-0.3Y ₂ O ₃	21-9-6/Y ₂ O ₃
2	Fe-21Cr-9Mn-6Ni-Si-0.15N-0.08C-0.4Ti-0.3ZrO ₂	21-9-6/ZrO ₂
3	Fe-21Cr-9Mn-6Ni-Si-0.15N-0.08C-0.4Ti-0.15Y ₂ O ₃ -0.15ZrO ₂	21-9-6/Y ₂ O ₃ -ZrO ₂

The three ODS steel compositions were properly mixed in a high-energy planetary ball mill (Fristch GmbH, Germany) for 20 h using mechanical alloying (MA). The ball mill operated at 300 min⁻¹ and was equipped with a stainless steel vial and balls of a 10-mm diameter. The MA process was conducted at room temperature under an argon protective atmosphere and a ball-to-powder weight ratio of 10:1 maintained in a toluene medium. The as-milled powders were hot pressed (VB ceramics Pvt. Ltd, India) with a pressure of 50 MPa at 1170 °C for 60 min under a vacuum of 10⁻² torr.

The grain distribution and oxide precipitates were investigated with a high-resolution transmission-electron microscope with energy-dispersive x-ray spectroscopy (HRTEM-EDX) (JEM/2100) operating at 200 kV. Thin-foil specimens of the ODS steel were mechanically thinned to 30 μm and punched into 3-mm discs. Dimpling (GATAN 656) followed by ion milling (GATAN 691) were applied before the TEM studies.

Uni-axial tensile tests were carried out at different temperatures of (301, 573 and 873) K, with a strain rate of 1 mm/min, using a universal testing machine (Instron 1362). The test specimens were machined according to

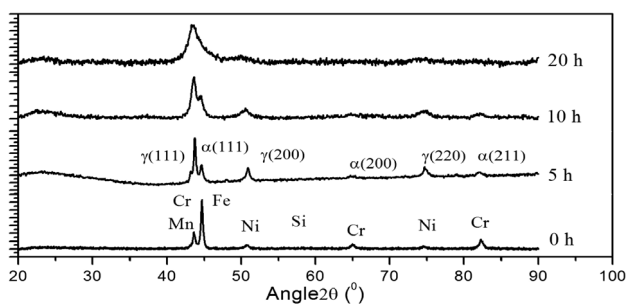


Figure 1: XRD pattern of 21-9-6/Y₂O₃-ZrO₂ powder

the ASTM-E8 standard, having a gauge length of 15 mm. The fracture-surface morphology of the tensile specimens was characterised using a high-resolution scanning-electron microscope (FEI Quanta FEG 200 HR-SEM).

3 RESULTS

3.1 Mechanical alloying

The XRD examination of the 21-9-6/Y₂O₃-ZrO₂ elemental composition is represented in **Figure 1**. The elemental composition at 0 h of MA showed individual elemental peaks of Fe, Cr, Mn and Ni. As observed, all the elemental peaks were transformed into a dual-phase structure of austenite (γ) and ferrite (α) phases after 5 h of MA, indicating a dissolution of the elemental powders into the base matrix. After 20 h of milling, the complete structure of the γ phase was observed. The MA process was optimized based on the transition rate between α and γ. The other two ODS compositions also follow similar changes.

3.2 Micro-structural analysis

The density of the three hot-pressed ODS steels was above 95 % of the theoretical density. The TEM microstructures of the 21-9-6/Y₂O₃ and 21-9-6/ZrO₂ steels are shown in **Figures 2a** and **2b**. According to the EDX data, the grains from the figures exhibit an elemental composition (w/%) of Fe, Cr, Mn and Ni.

Two different sizes of the grains, namely, a large grain (below 1 μm) and a small grain (below 200 nm) were observed in the bimodal distribution of the grains in the microstructure of the 21-9-6/Y₂O₃-ZrO₂ steel (**Figure 2c**). According to EDX data (**Figure 2d**), the location (+) of smaller grains contains the main elements of Zr, Y, Ti, Si and O.

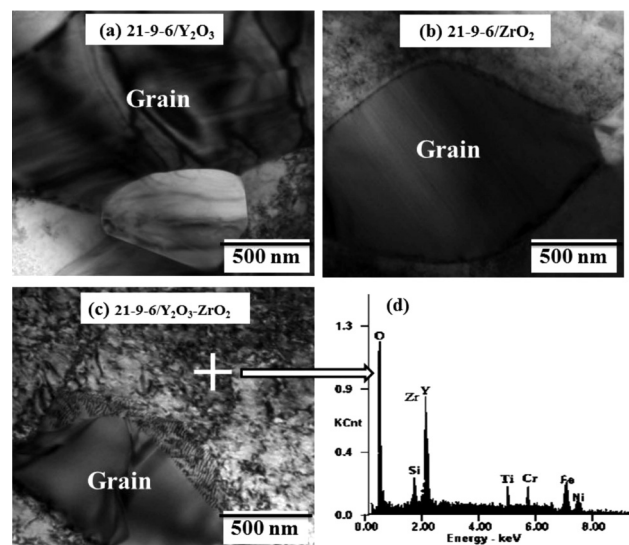


Figure 2: TEM microstructures of ODS steels

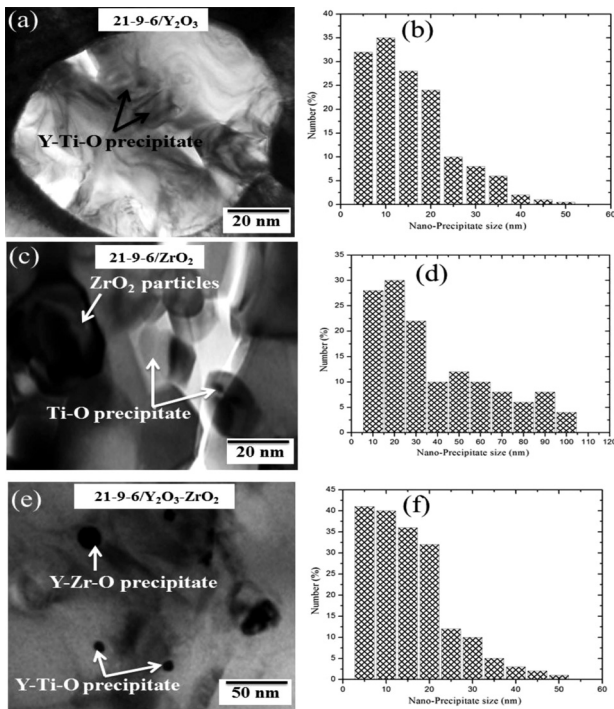


Figure 3: Appearance of nano-oxide precipitates

In order to further study the nano-oxide precipitates, higher-magnification TEM images of the oxide precipitates and size distribution (Figure 3) were obtained. The fine oxide precipitates of the Y-Ti-O rich composition of the microstructure of the 21-9-6/ Y_2O_3 steel vary from 5 to 50 nm according to the analysis of the particle-size frequency distribution (Figures 3a and 3b). On the other hand, the analysis of the ZrO_2 dispersed ODS steel (21-9-6/ ZrO_2) revealed coarse oxide particles of the ZrO_2 composition (Figure 3c) and fine precipitates of the Ti-O composition; a large variation in the size of the precipitates can be observed in Figure 3d. In the 21-9-6/ Y_2O_3 - ZrO_2 steel (Figure 3e), the precipitates mainly have the Y-Ti-O and Y-Zr-O compositions. The size of most of the oxide particles is in a range of 10–20 nm (Figure 3f).

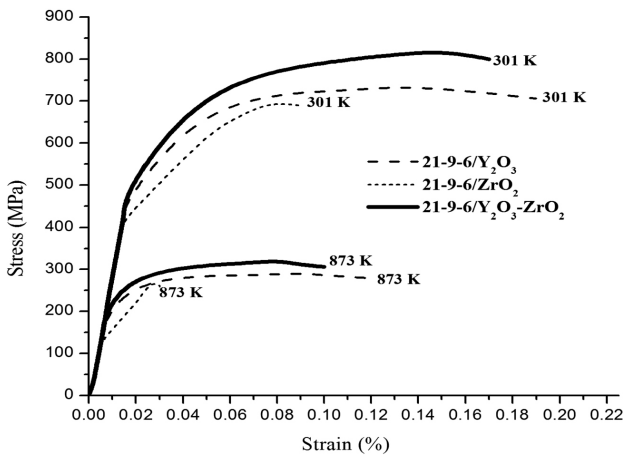


Figure 4: Stress-strain curves for hot-pressed ODS steels

The properties of the nano-oxide precipitates obtained for the three ODS steels are listed in Table 2. The calculations were performed on a large data array (a minimum of 1000 nano-sized particles per composition). As it is seen, the 21-9-6/ Y_2O_3 - ZrO_2 ODS steel exhibited the minimum size and the maximum density of the precipitates.

Table 2: Characteristics of nano-oxide precipitates

ODS steel	Density (per cc)	Size (nm)
21-9-6/ Y_2O_3	3.2×10^{15}	14
21-9-6/ ZrO_2	0.9×10^{15}	26
21-9-6/ Y_2O_3 - ZrO_2	5.6×10^{15}	12

3.3 Mechanical properties

3.3.1 Tensile properties

Figure 4 shows the stress-strain curves for three ODS steels at different temperatures. Five replicate tensile tests were performed for each ODS steel at each temperature. The yield strength (YS), tensile strength (UTS) and percent elongation (PE) of the three ODS steels at (301, 573 and 873) K are summarized in Table 3. At all the testing temperatures, the 21-9-6/ Y_2O_3 - ZrO_2 specimen exhibited higher YS and UTS values compared to the other ODS steels, due to the bimodal distribution of powder grains in the microstructure (Figure 2c) and

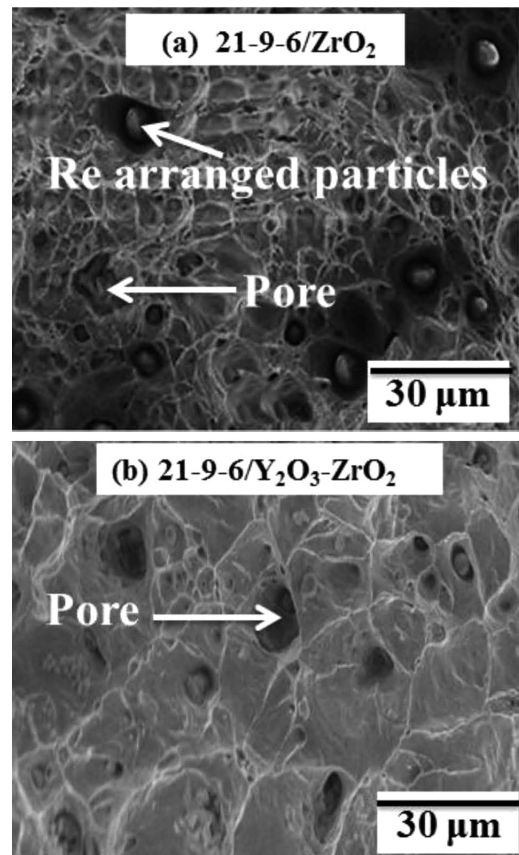


Figure 5: Fracture-surface morphology

Table 3: Mechanical properties of ODS steels at different test temperatures

Composition	301 K			573 K			873 K		
	YS (MPa)	UTS (MPa)	PE (%)	YS (MPa)	UTS (MPa)	PE (%)	YS (MPa)	UTS (MPa)	PE (%)
21-9-6/Y ₂ O ₃	466	734	19	421	667	17	181	291	12
21-9-6/ZrO ₂	419	696	9	374	626	6	132	269	3
21-9-6/Y ₂ O ₃ -ZrO ₂	494	820	17	452	756	15	199	321	10

Table 4: Comparison of microstructures and tensile strengths of different ODS austenitic steels

Composition	Process	Nano-oxide precipitate (nm)	UTS at 301 K (MPa)	UTS at 873 K (MPa)	Ref.
18Cr-8Ni-1Mo-0.5Ti-0.35Y ₂ O ₃	HIP	Y-Ti-Si-O (20)	1000	500	4
SS 316 +0.3 Ti + 0.35 Y ₂ O ₃	HIP	YAH (20)	729	459	10
16Cr-13Ni-0.2 Y ₂ O ₃ -0.45Zr	HIP+HR	—	970	370	11
14Cr-Zr	HIP	Y-Zr-O (8)	994	358	12
SS 304+0.35Y ₂ O ₃ +0.5Ti	HIP	Y-Ti-O (13)	940	415	13
SS 304 +0.3 Ti + 0.35Y ₂ O ₃	HIP	Y-Al-O (20)	1000	—	14
21-9-6/Y ₂ O ₃	HP	Y-Ti-O (14)	734	291	Present work
21-9-6/Y ₂ O ₃ -ZrO ₂	HP	Y-Ti-O & Y-Zr-O (12)	820	321	Present work

also due to the presence of nano-sized Y-Zr-O oxide precipitates (**Figure 3e**), which pinned the dislocations and retarded the grain growth.⁹

However, the PE values for the 21-9-6/Y₂O₃-ZrO₂ steel were slightly reduced when compared to the 21-9-6/Y₂O₃ steel at all the testing temperatures due to the high density of the nano-oxide precipitates (**Table 2**), making the material harder.¹⁰

3.3.2. Fractographic examination

The fracture surface of 21-9-6/ZrO₂ (**Figure 5a**) contained rearranged particles in some dimples. This indicated that failure occurred only at the particle interface, indicating a brittle failure. Because of this, the specimen exhibited poor mechanical properties (**Table 3**). However, the 21-9-6/Y₂O₃-ZrO₂ steel (**Figure 5b**) clearly showed a ductile fracture evidenced by fine dimples over the entire surface. This was also validated by the tensile properties (**Figure 4** and **Table 3**) of the specimen, showing the highest elongation among the ZrO₂ dispersed ODS steels.

4 DISCUSSION

4.1 Microstructure

The microstructure of the 21Cr-9Mn-6Ni ODS steel was studied based on the additions of different ODS materials (Y₂O₃ and ZrO₂). The microstructure of the Y₂O₃ ODS steel has the appearance of Y-Ti-O nano-oxide precipitates, indicating the presence of Y₂O₃ dissolved in the austenite matrix (**Figure 3a**). It is then combined with Ti to form nano-oxide precipitates of Y-Ti-O in the ODS steel.⁸

The microstructure of the ZrO₂ ODS steel contains ZrO₂ particles (**Figure 3c**) indicating that it does not dissolve in an austenite lattice and may not form a solid

solution with Ti. However, in the cases of Y₂O₃ and ZrO₂ added to steel, the ZrO₂ particles are dissolved in the austenite matrix, forming a thermodynamically favourable oxide precipitate with Y₂O₃ (**Figure 3e**). Similar observations^{8,11,12} were reported for the behaviour of ZrO₂ during the consolidation of ODS steel.

4.2 Mechanical properties

The mechanical properties of the 21Cr-9Mn-6Ni ODS steel were studied as a function of different ODS materials (Y₂O₃ and ZrO₂) added.

The Y₂O₃ ODS steel from the present work exhibited a lower UTS value (**Table 4**) compared to the other ODS austenitic steels developed previously. This is due to the HIP^{4,10,13,14} consolidation process used for developing ODS steels, which is an effective process compared to the HP process used in this study.

The Y₂O₃ and ZrO₂ ODS steels from the present work can have the tensile-strength values comparable to the other ODS steels developed previously.¹⁰⁻¹⁴ The bimodal distribution of the grains (**Figure 2e**) and the formation of the Y-Ti-O and Y-Zr-O nano-oxide precipitates (**Table 2** and **Figure 3e**) in the microstructure explain the high tensile-strength values of the ODS 21Cr-9Mn-6Ni steel.

5 CONCLUSIONS

In this research, the authors investigated the microstructure and mechanical properties of Y₂O₃ and ZrO₂ 21Cr-9Mn-6Ni-0.4Ti ODS steels. Three ODS steels (21-9-6/Y₂O₃, 21-9-6/ZrO₂, 21-9-6/Y₂O₃-ZrO₂) were made from elemental powders and consolidated by hot pressing. The results were compared and the conclusions are summarized as follows:

According to the TEM results, a bimodal distribution of the grains was found in the microstructure of the

21-9-6/Y₂O₃-ZrO₂ ODS steel, which indicates that ZrO₂ was decomposed and dissolved into the austenite matrix only in the presence of Y₂O₃.

The nano-oxide precipitates in the 21-9-6/Y₂O₃-ZrO₂ steel have a smaller size and a higher density compared to the other two ODS steels. This is due to the formation of Y-Zr-O and Y-Ti-O oxide precipitates with the additions of Y₂O₃ and ZrO₂ particles.

The composition of the 21-9-6/Y₂O₃-ZrO₂ ODS steel showed higher yield and tensile strengths compared to the other two ODS steels studied. This is due to the homogeneous dispersion of the nano-oxide particles and bimodal distribution of powder grains.

The fractography studies of the surface of the 21-9-6/Y₂O₃-ZrO₂ ODS steel revealed that the presence of regular-shaped dimples indicates a ductile failure.

Based on the above findings, it can be concluded that an addition of Y₂O₃ and ZrO₂ facilitates attractive mechanical properties of 21-9-6 steels.

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